

THE UNIVERSITY *of* EDINBURGH  
School of Chemistry

SNUG PG NMR course, Edinburgh

2<sup>nd</sup> December 2024

There's more to NMR  
than just protons









Dr Andrew Hall

# NMR periodic table



H																					He
Li	Be											B	C	N	O	F					Ne
Na	Mg											Al	Si	P	S	Cl					Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br					Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I					Xe
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At					Rn
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts					Og
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu							
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr							

**Nuclear spin:**

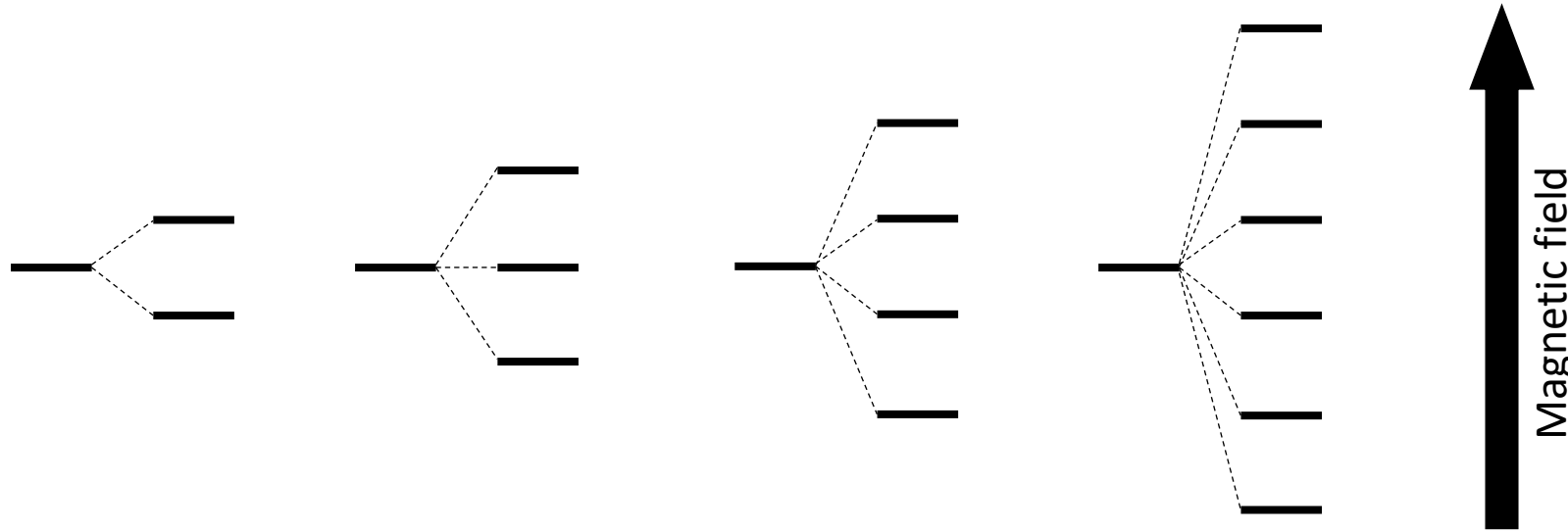
	1/2		3/2		7/2		5
	1		5/2		9/2		8

# Spin



Number of energy levels:

$$2i + 1$$

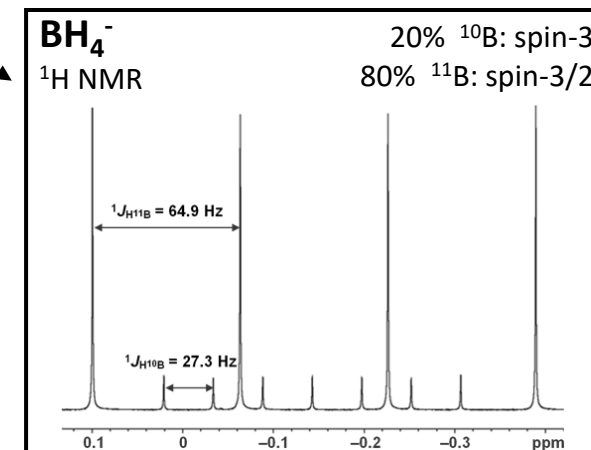
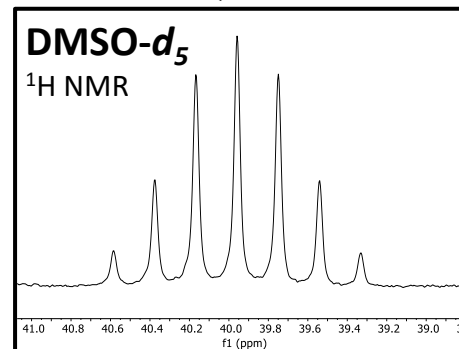
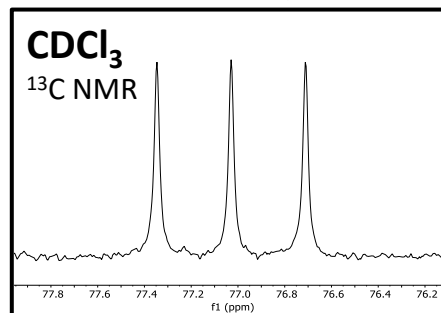


Spin-1/2  
<sup>1</sup>H, <sup>13</sup>C, <sup>19</sup>F, <sup>31</sup>P...

Spin-1  
<sup>2</sup>H, <sup>6</sup>Li, <sup>14</sup>N

Spin-3/2  
<sup>7</sup>Li, <sup>11</sup>B, <sup>23</sup>Na <sup>35</sup>Cl...

Spin-5/2  
<sup>17</sup>O, <sup>27</sup>Al, <sup>23</sup>Na <sup>35</sup>Cl...



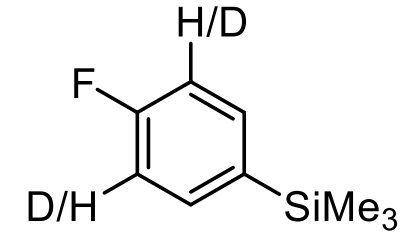
# Quadrupolar nuclei



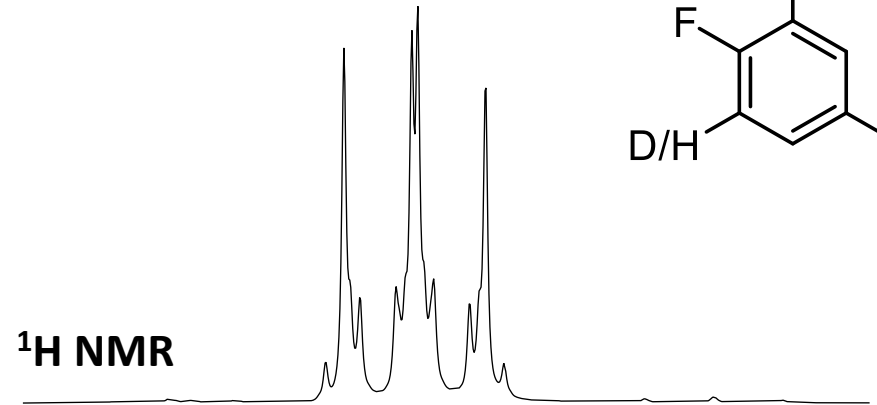
Nuclei with a spin quantum number larger than  $\frac{1}{2}$  are **quadrupolar**.

Quadrupolar nuclei have an **asymmetrical** charge distribution.

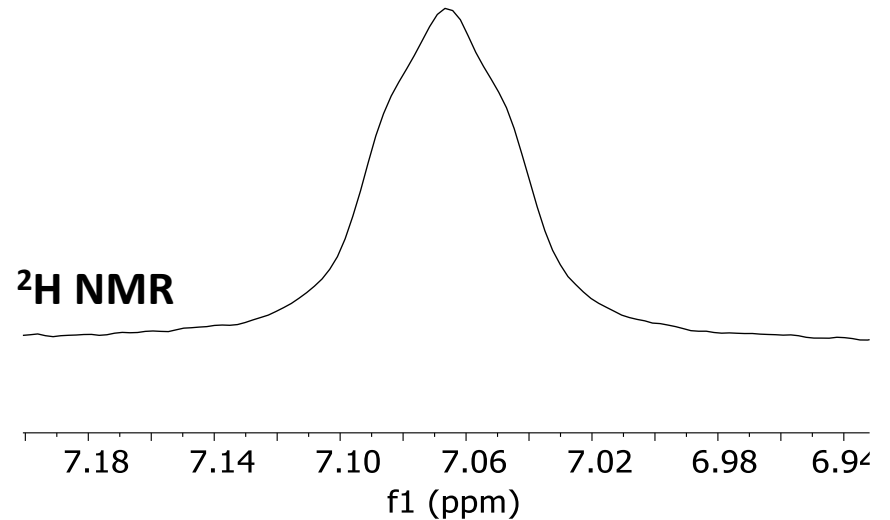
This causes faster relaxation, resulting in broad peaks.



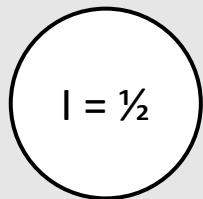
$^1\text{H NMR}$



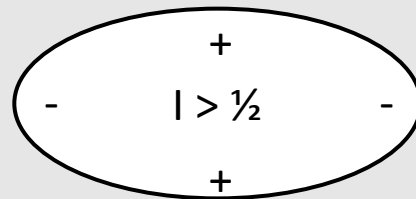
$^2\text{H NMR}$



Charge distribution:



Isotropic

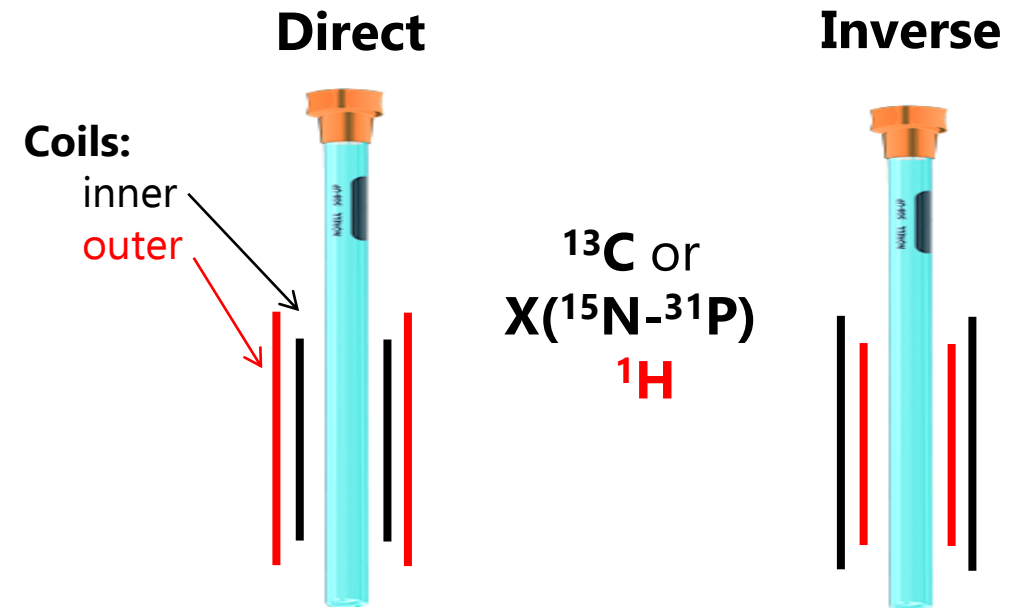
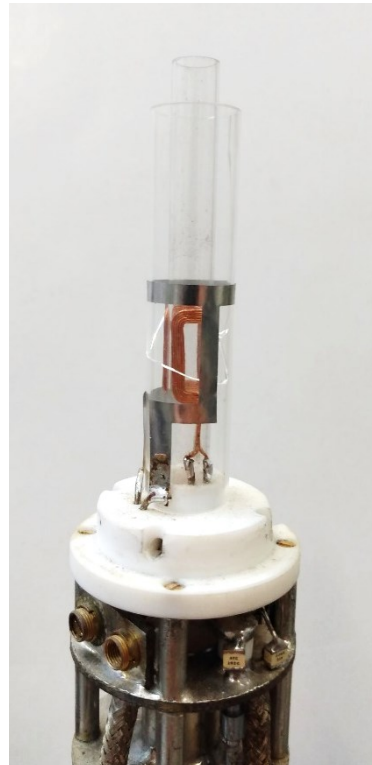
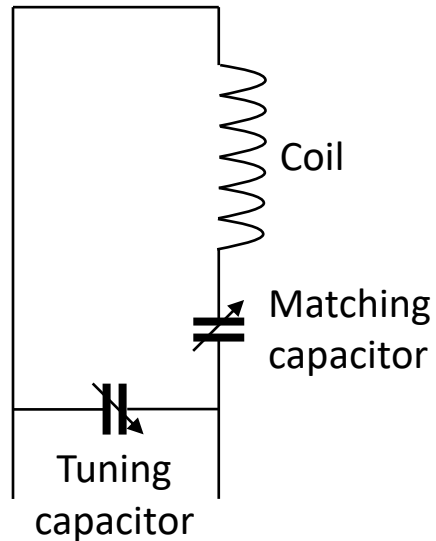


Quadrupolar

# Probes



NMR probes usually have two coils, one for  $^1\text{H}$  (and  $^{19}\text{F}$ ) and a second that can be tuned to a range of nuclei. Depending on how the probe is constructed, the coils can be optimized for different nuclei. Cooling the probe to low temperature (77 or 20 K) can increase the sensitivity.



# NMR periodic table



Most probes only detect nuclei with  $\gamma > \gamma_N$

H																	He																														
Li	Be											B	C	N	O	F	Ne																														
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<table border="1"> <tr> <td>La</td> <td>Ce</td> <td>Pr</td> <td>Nd</td> <td>Pm</td> <td>Sm</td> <td>Eu</td> <td>Gd</td> <td>Tb</td> <td>Dy</td> <td>Ho</td> <td>Er</td> <td>Tm</td> <td>Yb</td> <td>Lu</td> </tr> <tr> <td>Ac</td> <td>Th</td> <td>Pa</td> <td>U</td> <td>Np</td> <td>Pu</td> <td>Am</td> <td>Cm</td> <td>Bk</td> <td>Cf</td> <td>Es</td> <td>Fm</td> <td>Md</td> <td>No</td> <td>Lr</td> </tr> </table>																		La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr
La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu																																	
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**Nuclear spin:**

<span style="color: red;">■</span>	1/2	<span style="color: yellow;">■</span>	3/2	<span style="color: cyan;">■</span>	7/2	<span style="color: gray;">■</span>	5
<span style="color: orange;">■</span>	1	<span style="color: green;">■</span>	5/2	<span style="color: magenta;">■</span>	9/2	<span style="color: darkgray;">■</span>	8

# Natural abundance



For many heteronuclei, the NMR-active isotope only makes up a small fraction of all nuclei at natural abundance. Some elements have multiple NMR-active isotopes.

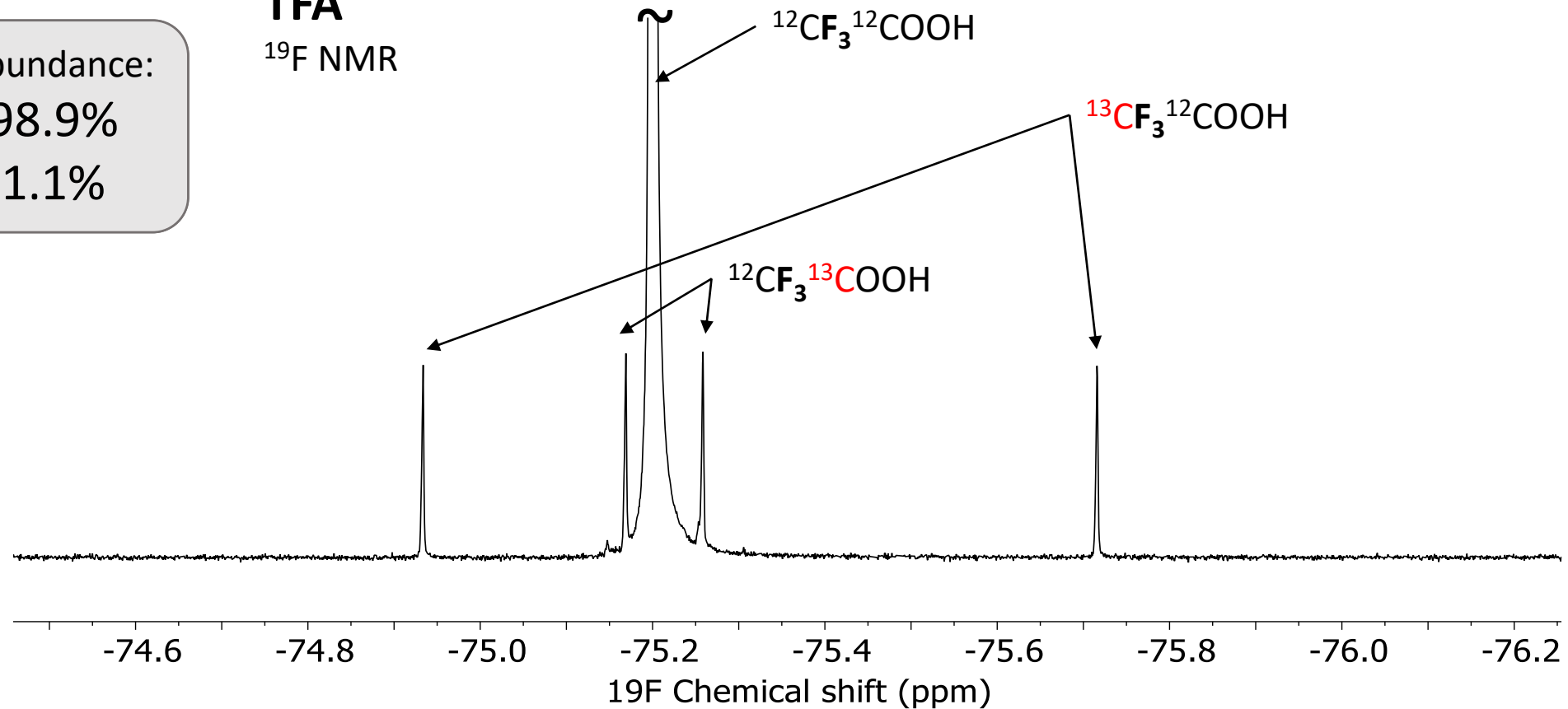
Natural abundance:

$^{12}\text{C} = 98.9\%$

$^{13}\text{C} = 1.1\%$

**TFA**

$^{19}\text{F}$  NMR

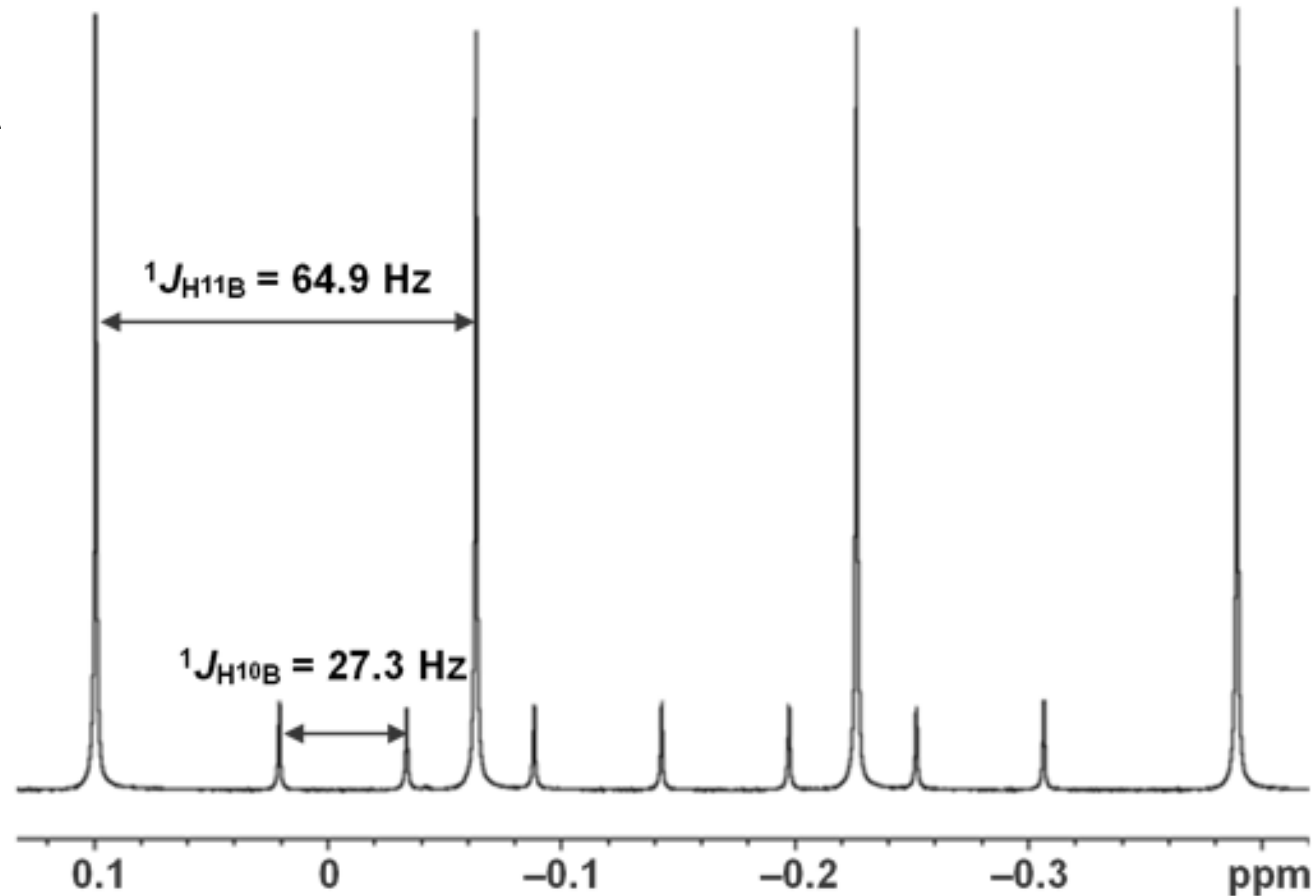


# Natural abundance



For many heteronuclei, the NMR-active isotope only makes up a small fraction of all nuclei at natural abundance. Some elements have multiple NMR-active isotopes.

**BH<sub>4</sub><sup>-</sup>**  
<sup>1</sup>H NMR



Natural abundance:  
<sup>10</sup>B = 20%, spin-3  
<sup>11</sup>B = 80%, spin-3/2



# NMR periodic table



Most probes only detect nuclei with  $\gamma > \gamma_N$

Some nuclei have too low natural abundance to detect

H																			He																												
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La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu																																	
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# NMR periodic table



Most probes only detect nuclei with  $\gamma > \gamma_N$

Some nuclei have too low natural abundance to detect

Common heteronuclei include  $^7\text{Li}$ ,  $^{10/11}\text{B}$ ,  $^{13}\text{C}$ ,  $^{15}\text{N}$ ,  $^{19}\text{F}$ ,  $^{27}\text{Al}$ ,  $^{29}\text{Si}$ ,  $^{31}\text{P}$  and  $^{119}\text{Sn}$

H																			He
Li	Be											B	C	N	O	F			Ne
Na	Mg											Al	Si	P	S	Cl			Ar
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br			Kr
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I			Xe
Cs	Ba		Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At			Rn
Fr	Ra		Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Nh	Fl	Mc	Lv	Ts			Og

La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

**Nuclear spin:**

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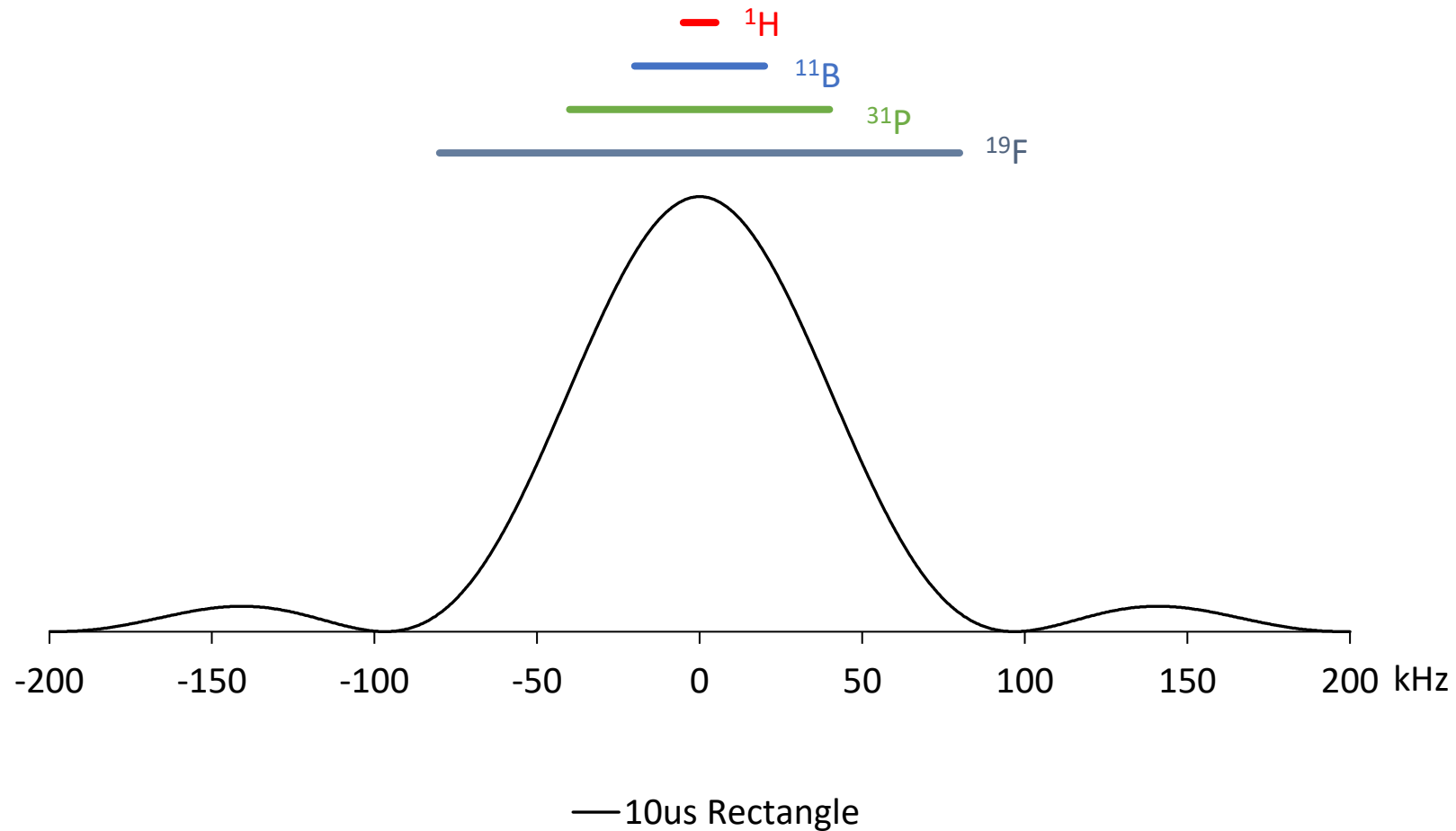
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# Acquisition

# Excitation width



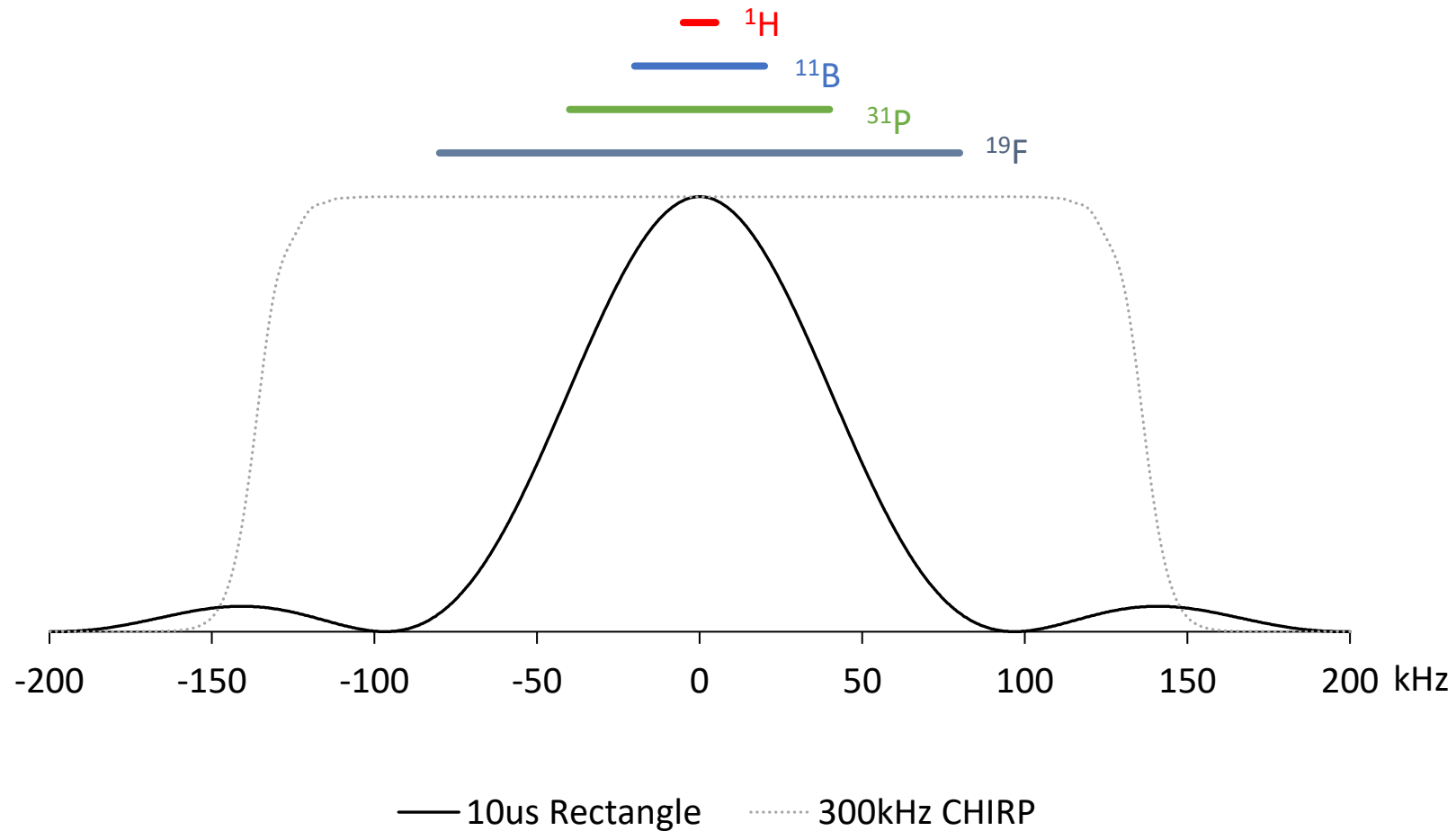
Standard pulses (typically  $\sim 10 \mu\text{s}$  duration) may not excite all peaks equally for nuclei with wide spectral ranges.



# Excitation width



Standard pulses (typically  $\sim 10 \mu\text{s}$  duration) may not excite all peaks equally for nuclei with wide spectral ranges. Other pulses can offer more uniform excitation profiles.

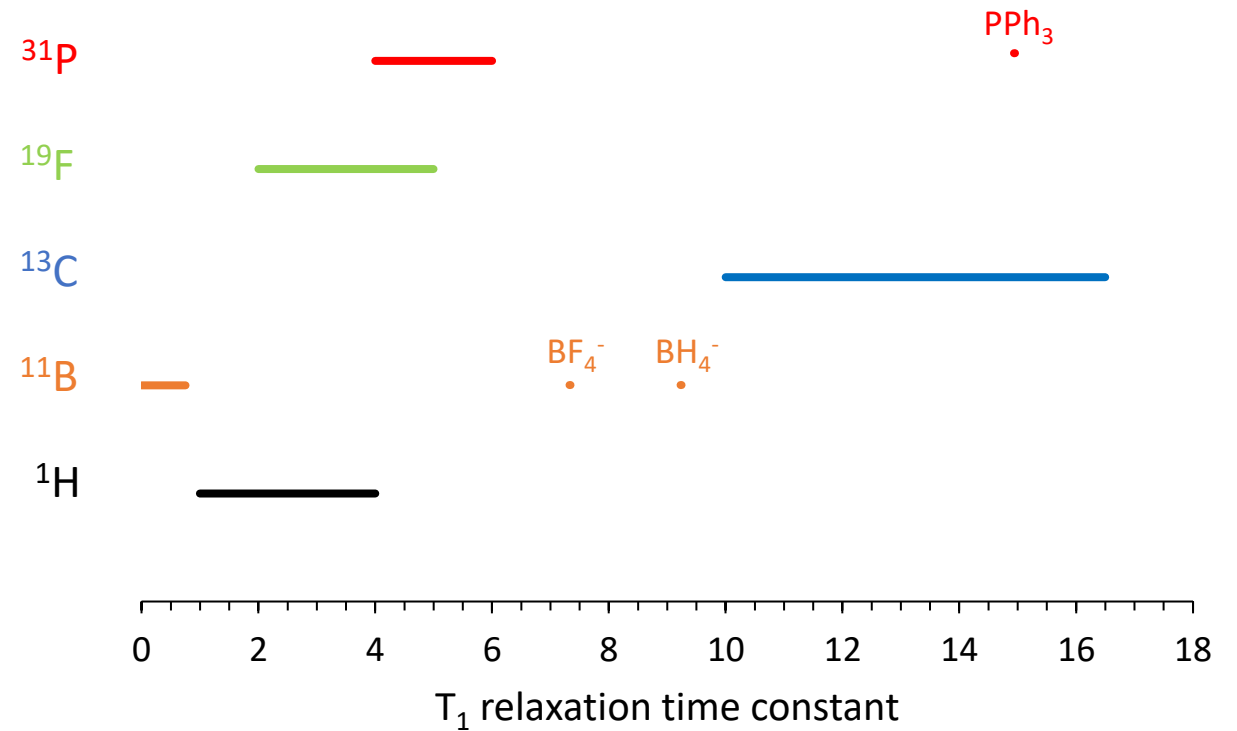
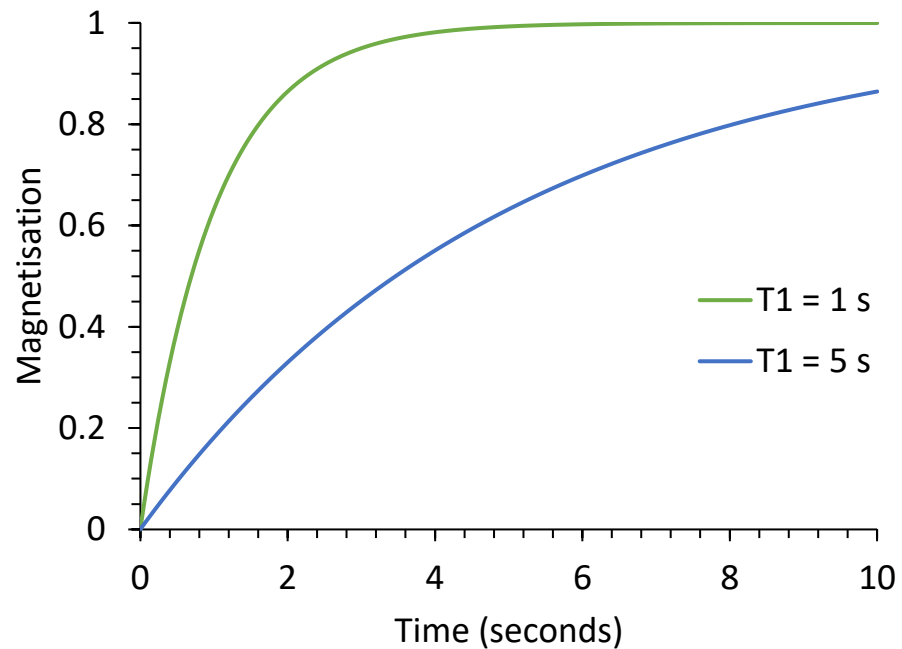


# Relaxation time



After the pulse, the NMR signal takes several seconds to recover.

The time taken for the signal to recover is determined by the  $T_1$  relaxation time constant for the nucleus.



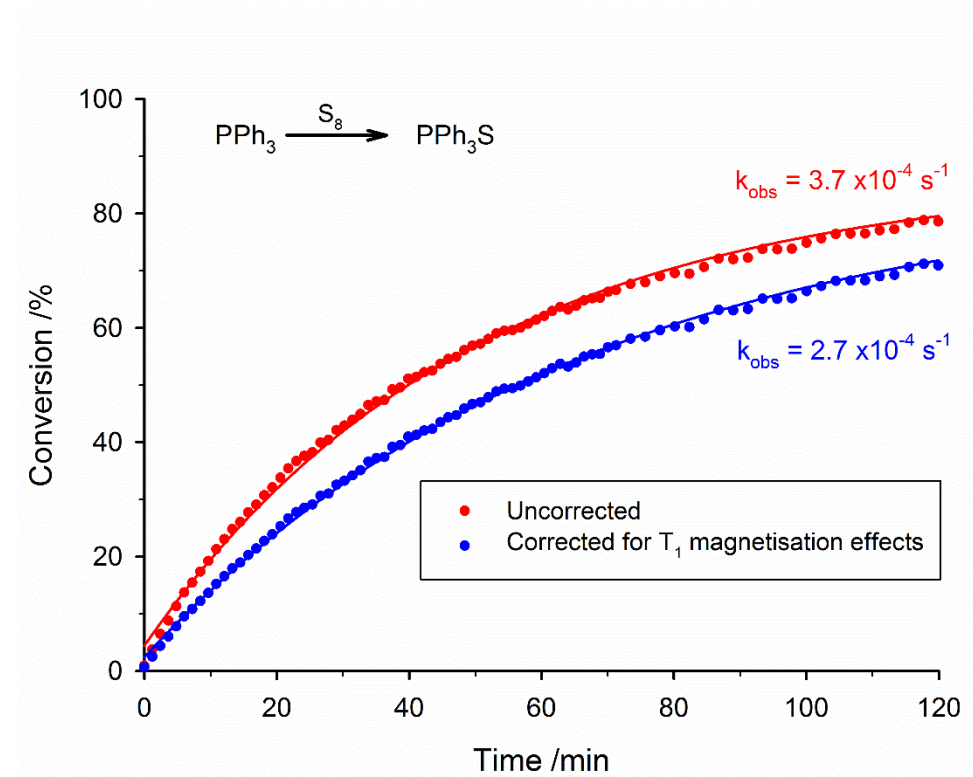
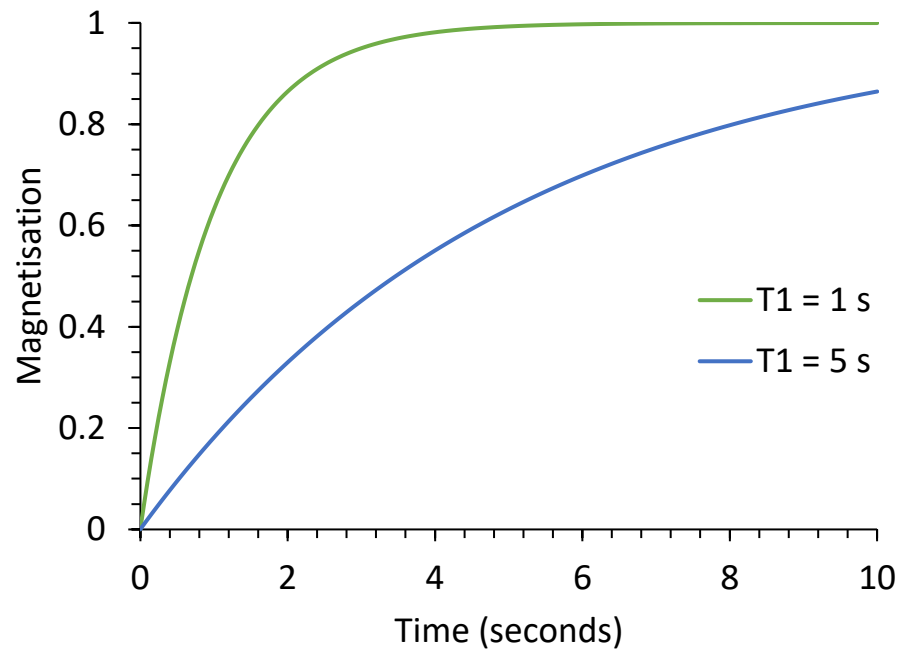
# Relaxation time



After the pulse, the NMR signal takes several seconds to recover.

The time taken for the signal to recover is determined by the  $T_1$  relaxation time constant for the nucleus.

The spectrum will only be quantitative if the signal is allowed to fully recover, usually  $>5 \times T_1$ .

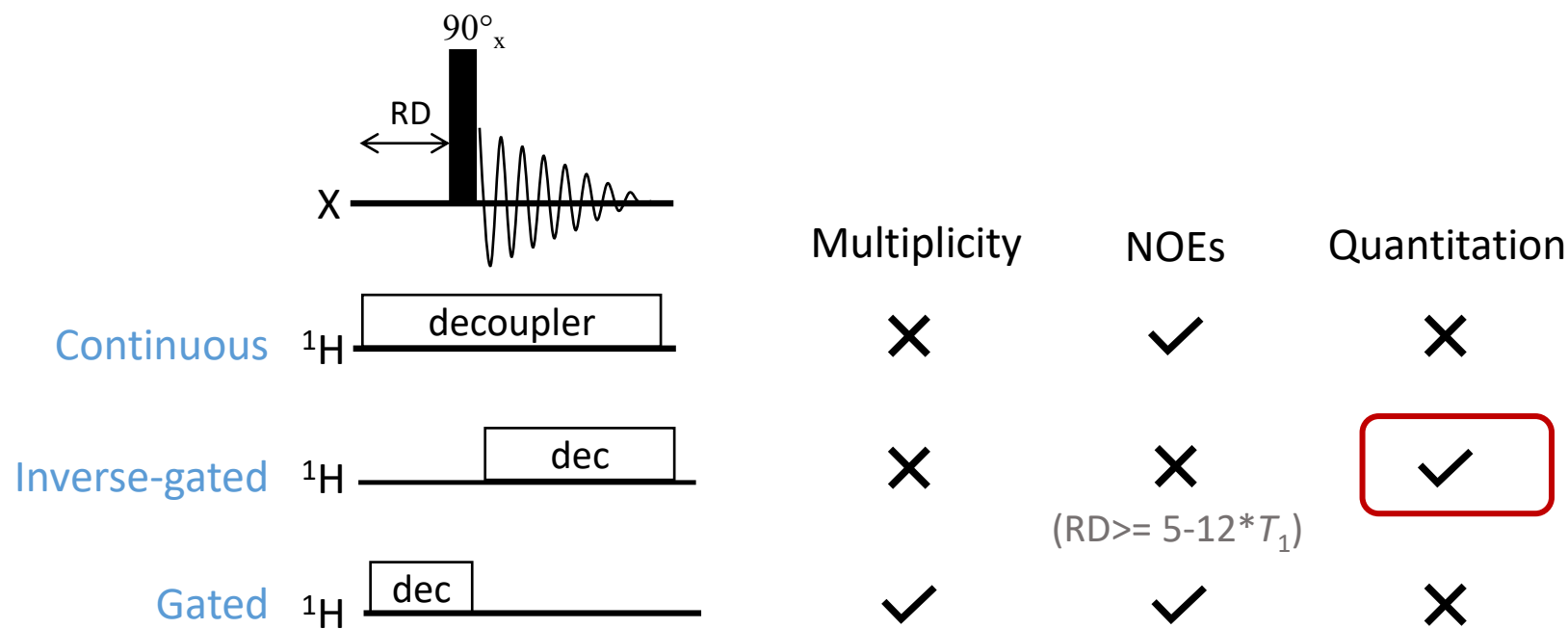


*Catal. Sci. Technol.*, 2016,6, 8406-8417

# Decoupling



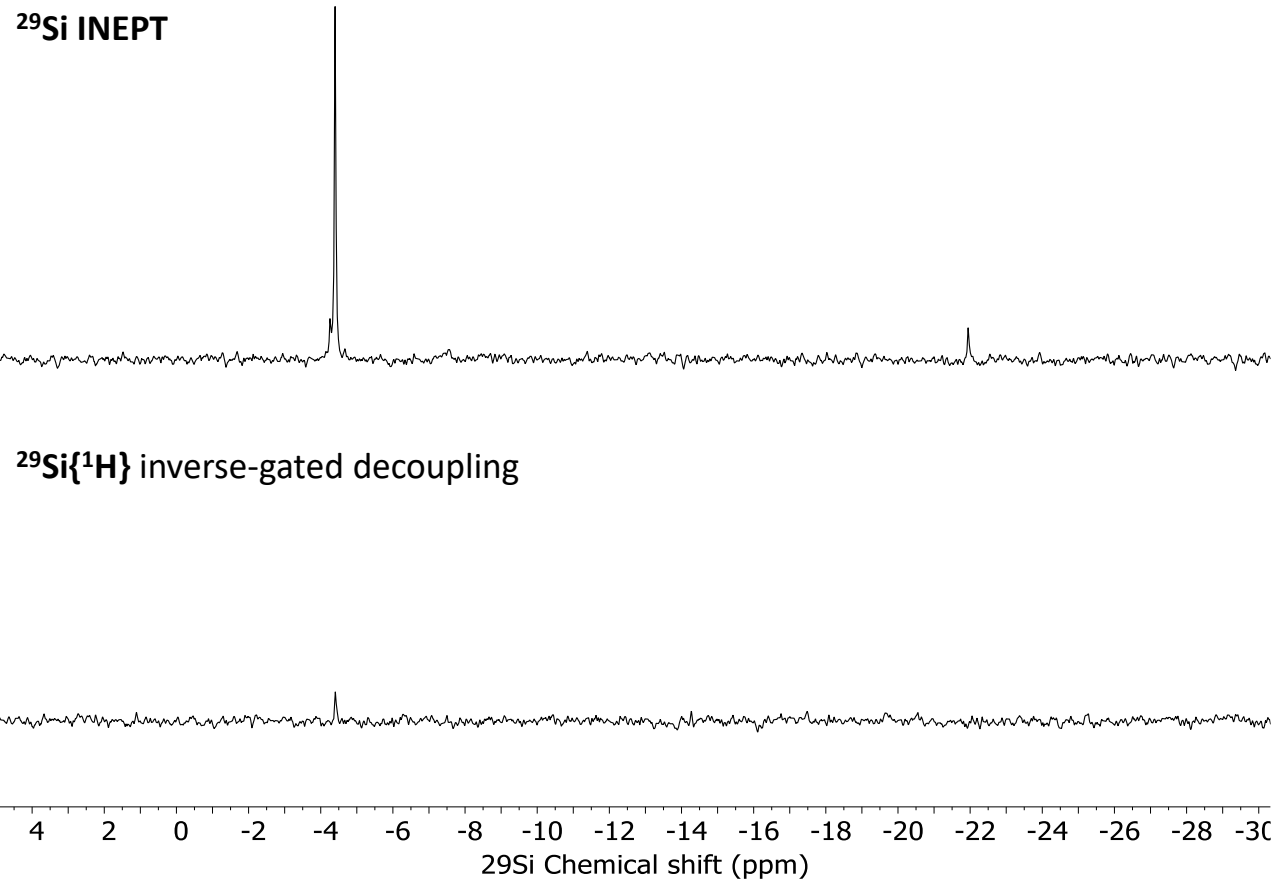
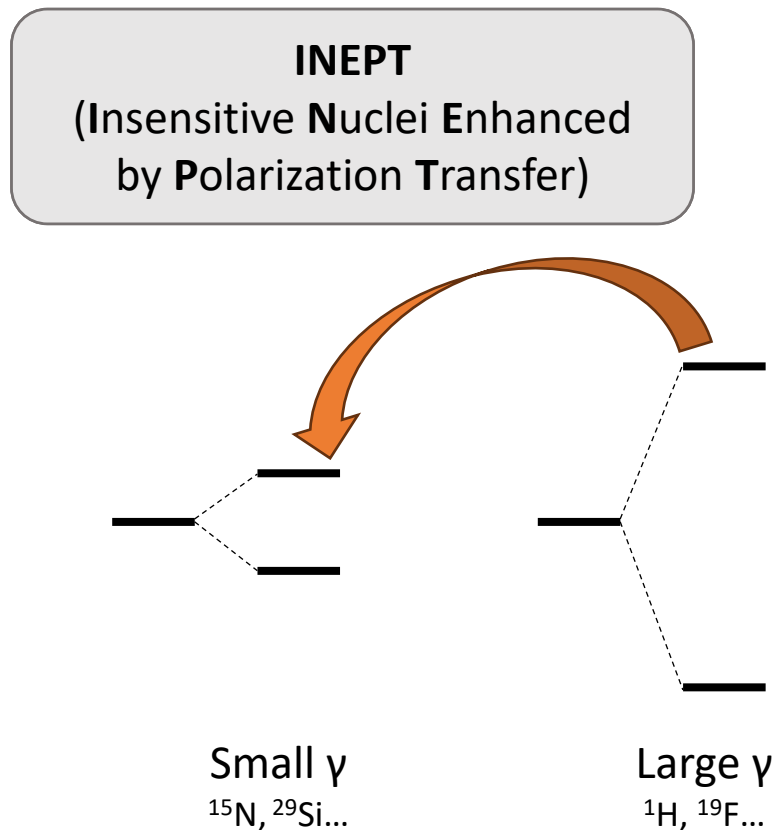
Decoupling uses the second coil to saturate the energy levels of the coupled nucleus, removing peak splitting. Different decoupling schemes can either enhance signal or enable quantitation.



NOEs (Nuclear Overhauser Effects)



Nuclei with a low gyromagnetic ratio (e.g.  $^{14}\text{N}$  or  $^{29}\text{Si}$ ) have lower levels of polarization, resulting in weak signals. Signals can be enhanced by transferring polarization from nearby nuclei with higher gyromagnetic ratio (e.g.  $^1\text{H}$  or  $^{19}\text{F}$ )





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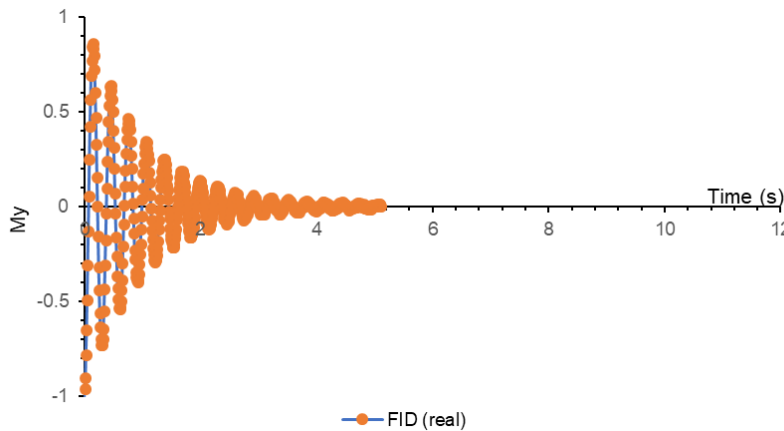
# Processing

# Zero filling

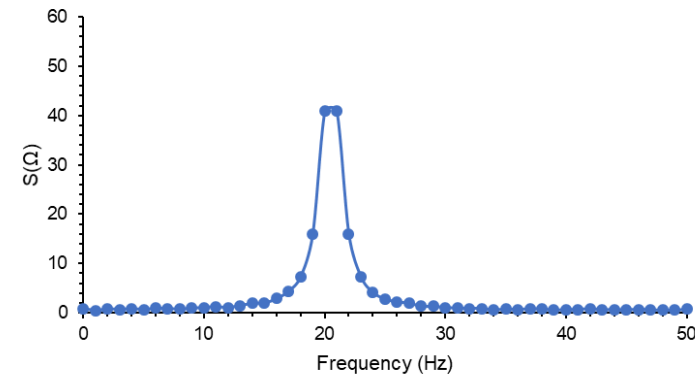


Zero filling gains additional frequency resolution for free, by increasing the number of points available for the Fourier transform. There is no new information added, but the extra points can help with integrating and fitting peaks.

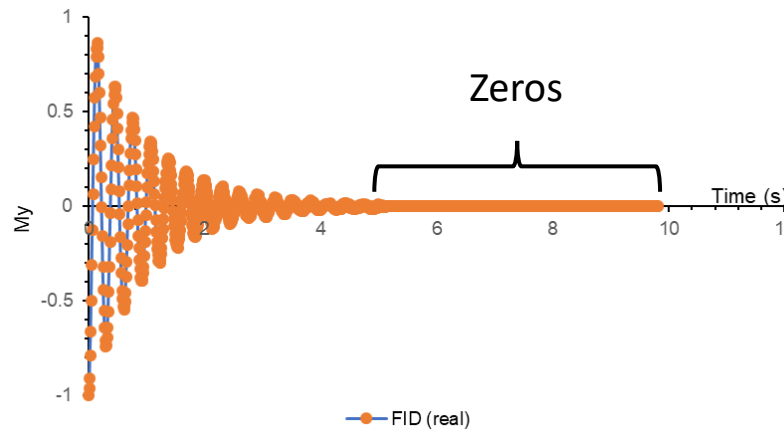
Normal data



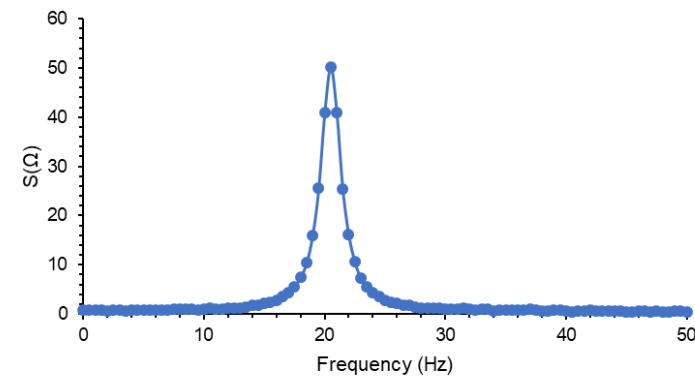
Frequency domain spectrum



Zero filled



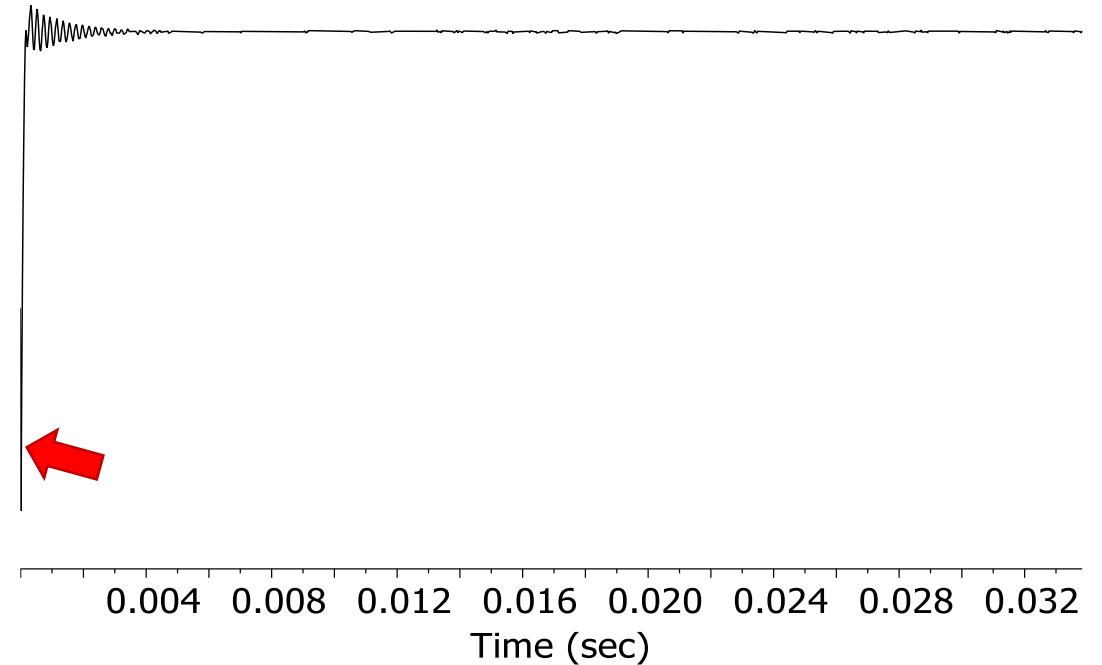
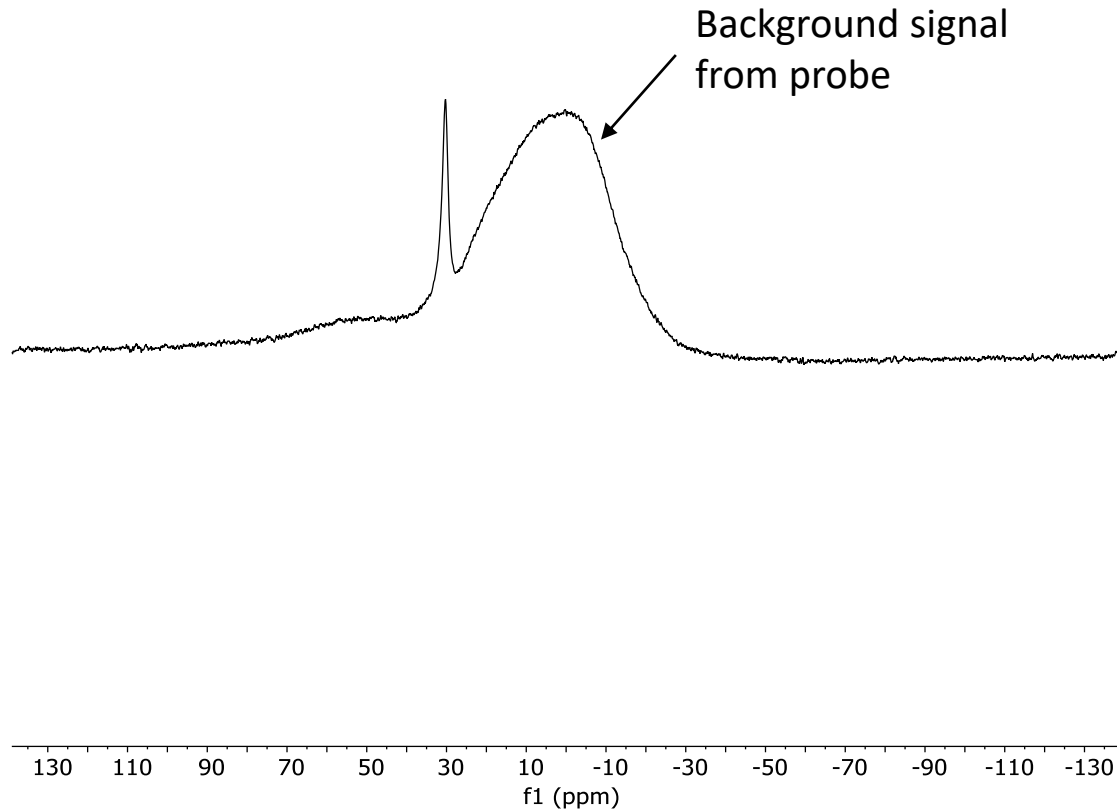
Frequency domain spectrum



# Background signals



Fast decaying signals from materials in the probe/glass result in broad background signals.



# Background signals



Fast decaying signals from materials in the probe/glass result in broad background signals.  
Removing the first few points from the FID and then using backward linear prediction can restore a flat baseline.

